Year 9 Science	Emerging	Developing	Secure	Mastery
		Term 1		
	Know that genetic information	Describe chromosomes	Explain the relationship between	Explore the role of
	is found in the nucleus of a	and their role in	chromosomes, genes and DNA.	scientists in the discovery
	cell.	transferring heredity		of DNA and evaluate the
		information to offspring.	Explain why offspring of the same	relative importance of their
	Know that offspring get half		parents may look very different.	contributions.
	their genetic information from	Describe cloning as one		
	their Mum and half from their	parent producing new	Explain how artificial cloning is	Explore and evaluate the
	Dad.	individuals and identify	performed – for example in the	advantages and
		examples of cloning that	creation of Dolly the sheep.	disadvantages of artificial
	Understand that clones are	occur naturally; describe		cloning; compare and
	genetically identical to their	natural cloning as asexual	Explain how every new individual	contrast asexual and
	parent.	reproduction.	produced by sexual reproduction	sexual reproduction.
B1 – Cells &			is genetically unique.	
Genetics	Identify an animal and a plant	Describe how fertilised egg		Explain the impact of slight
	cell.	cells contain half of the	Describe the functions of the	'changes' to DNA passed
		chromosomes from each	nucleus, cell membrane,	on from parents to
	Recognise that substances	parent with a random mix	mitochondria, cytoplasm, cell	offspring.
	are able to move in and out of	of genetic information from	wall, vacuole and chloroplast.	
	cells.	each parent.		Explain how different
			Describe the process of diffusion,	structures help organisms
		Recognise and label	and name the materials needed	to survive.
		normal and specialised	by and those removed from the	
		animal and plant cells.	cell.	
		Use a microscope to make		

observations.

Year 9 Science	Emerging	Developing	Secure	Mastery
		Recognise the role of diffusion in living organisms.		
	Understand that you can catch some diseases from others.	Describe and give examples of a way in which diseases are spread.	Describe several examples of how specific diseases are spread and suggest how their spread may be reduced.	Consider suggestions to reduce the spread of specific diseases and justify decisions.
	Know that your body can fight off disease.	Describe the body's mechanisms to prevent infection.	Describe the roles of white blood cells in fighting infection.	Explain why we rarely catch the same infectious
	Understand that bacteria need certain conditions to survive.	State examples of diseases caused by microbes.	Describe the characteristics of different microbes.	disease twice but may catch influenza over and over again.
B2 – Keeping Healthy	Know that antibiotics can be used to kill bacteria.	Describe the conditions that bacteria need to survive.	Compare bacterial growth in different parts of the home.	Evaluate a model of a type of microbe.
	Know that vaccines can prevent you from catching a disease.	Describe the effect of antibiotics on bacteria.	Explain how bacteria become resistant to antibiotics.	Analyse data about bacterial growth.
	Know that plants can get disease too.	Describe what a vaccine is and how vaccines were	Explain how vaccines prevent a viral infection.	Explain what superbugs are and evaluate their impact on society.
		discovered.	Describe the effects of plant diseases on plant growth.	Evaluate the risks of vaccines and disease.

Year 9 Science	Emerging	Developing	Secure	Mastery
		State the names of some		Explain how plant diseases
		plant diseases.		can be avoided.
	Understand that chemical reactions in the body release	Describe the purpose of respiration.	Describe and explain aerobic respiration using a word	Explain the role of respiration in building up
	energy.	respiration.	equation.	complex molecules.
	Identify what chemicals are needed for respiration. Understand that if you exercise with not enough oxygen your muscles will ache.	Define anaerobic respiration and give examples of sports that use anaerobic respiration.	Explain why some sports rely mainly on aerobic respiration while others require anaerobic respiration.  Describe and explain some evidence to show the products of	Describe and explain the effects on the body of anaerobic respiration and explain 'oxygen debt'  Plan an investigation to test a hypothesis about
B4 – Growth & Respiration	Know there are two types of cell growth.	Identify some living things that carry out anaerobic respiration and identify	anaerobic respiration and plan an investigation into fermentation.	anaerobic respiration, analyse the data and evaluate the investigation.
	Know there are two types of stem cells.	some applications.  State what mitosis and meiosis is.  State the names of the two types of stem cells.	Describe the process of mitosis and meiosis.  Describe the process of embryonic and adult stem cells.	Explain mitosis and meiosis using key terminology.  Explain the process of two types of stem cells, giving advantages and disadvantages for each.

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Year 9	Emerging	Developing	Secure	Mastery
Science	Emerging	Developing	Secure	Plastery

	Term 2					
	Know the three things an atom	Draw and label the	Draw, label and describe the	Explain the effects each		
	is made up of.	structure of the atom.	charges and mass of the three	type of radiation can have.		
			subatomic particles.			
	Know there is three types of	Name the three types of		Interpret data from a graph		
	radiation.	radiation.	Describe the properties of each	to work out half life and		
			type of radiation.	relate this to how		
	State what half-life is.	Draw a half-life graph.		dangerous it is.		
			Interpret data from a graph to			
	State some effects of	State the definition for	work out half-life.	Use a mathematical		
	radiation.	irradiation and		equation to prove isotopic		
		contamination.	Describe the effects of irradiation	abundance of elements to		
P5 - Radiation	State what an isotope is.		and contamination.	explain the numbers on		
r 3 - Naulation		State the definition for an		the periodic table.		
	Recognise that nuclear decay	isotope and give some	Use a mathematical equation to	Write decay equations for		
	happens.	examples.	prove isotopic abundance of	Alpha, Beta & Gamma		
		State what alpha and beta	elements.	decay.		
	Link theory of radiation decay	decay does to the nucleus				
	to the Chernobyl	of atom.	Write decay equations for Alpha	Link theory of radiation		
	documentary.		& Beta decay.	decay to the Chernobyl		
		Link theory of radiation		documentary.		
		decay to the Chernobyl	Link theory of radiation decay to			
		documentary.	the Chernobyl documentary.			

Year 9 Science	Emerging	Developing	Secure	Mastery
	Identify basic lab equipment.	Name and draw equipment	Select and draw apparatus	Use particle diagrams to
		and explain obvious	accurately; explain safety	explain the differences in
	Use laboratory equipment	laboratory risks.	precautions.	energy and forces between
	safely to gather evidence.			the particles in different
		Compare the properties of	Draw circle diagrams to	states of matter,
	Represent particles as circles.	solids, liquids and gases.	demonstrate the differences	accounting for differences
			between the arrangement of	in their properties.
	Label a diagram with correct	Use correct terminology	particles in solids, liquids and	
	changes of state.	and the particle model to	gases, and describe their	Use the particle model to
		describe changes of state,	different properties.	explain latent heat and
	Identify objects that may float	including evaporation.		how impurities affect
	or that may sink.		Interpret and explain data relating	melting and boiling points.
P6 – States of		Suggest why some objects	to melting and boiling points.	
Matter	State that hot objects give out	float and others sink.		Apply ideas about density
raccor	heat.		Use the concepts of density,	and upthrust to predict the
		Describe the transfer of	displacement and upthrust in	outcomes of various
		energy by heating and	explaining floating and sinking.	situations
		cooling.		
			Explain the relationship between	Compare the transfer of
			energy transfer and temperature	energy by conduction and
			difference.	by radiation.

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Year 9	Emorging	Dovoloning	Secure	Mastery
Science	Emerging	Developing	Secure	Plastery

	Term 3					
	Identify basic lab equipment.	Name and draw equipment	Select and draw apparatus	Use particle diagrams to		
		and explain obvious	accurately; explain safety	explain the differences in		
	Use laboratory equipment	laboratory risks.	precautions.	energy and forces between		
	safely to gather evidence.			the particles in different		
		Compare the properties of	Draw circle diagrams to	states of matter,		
	Represent particles as circles.	solids, liquids and gases.	demonstrate the differences	accounting for differences		
			between the arrangement of	in their properties.		
	Label a diagram with correct	Use correct terminology	particles in solids, liquids and			
	changes of state.	and the particle model to	gases, and describe their	Use the particle model to		
		describe changes of state,	different properties.	explain latent heat and		
	Understand that some	including evaporation.		how impurities affect		
	chemical reactions produce		Interpret and explain data relating	melting and boiling points.		
C1 – Climate	heat.	Describe what is meant by	to melting and boiling points.			
change		the terms exothermic and		Use energy-level diagrams		
	Know what gas is produced by	endothermic reactions,	Explain the energy changes taking	to compare the energy in		
	combustion.	with examples.	place during an exothermic and	the reactants and products		
			an endothermic reaction.	of exothermic and		
	State the four greenhouse	Describe what a catalyst is		endothermic reactions,		
	gases.	and give examples.	Interpret data to explain how a	explaining the energy		
			catalyst affects a reaction.	changes in the particles.		
	State the water sources in the	Describe features of				
	UK.	physical and chemical	Describe the effects of acid rain.	Explain, using an equation,		
		changes, recognising how		the effects of acid rain.		
		mass is conserved.	Describe effects of greenhouse			
			gases.			

Year 9 Science	Emerging	Developing	Secure	Mastery
		Describe how combustion	Describe how potable water is	Describe the effects of
		contributes to acid rain.	obtained.	greenhouse gases and how
				we can mitigate them.
		State the greenhouse		
		gases and where they		
		come from.		
		State the definition of		
		potable water.		
	Idontify books lob on viscont	Name and draw any inneres	Calact and draw amparatus	Define elemente use
	Identify basic lab equipment.	Name and draw equipment	Select and draw apparatus	Define elements, use
	Llas labaratam raguinmaant	and explain obvious	accurately; explain safety	symbols, link the
	Use laboratory equipment	laboratory risks.	precautions.	organisation of the Periodic Table to element
	safely to gather evidence.	Civo como ovemplos ef	Cive exemples of elements and	features.
	Cive same examples of	Give some examples of	Give examples of elements and	reatures.
	Give some examples of elements.	elements, locate them in the Periodic Table and use	explain how they are organised in the Periodic Table.	Salast and justify the use
	eterrients.		the Periodic Table.	Select and justify the use of elements for different
C2 - Chemical	Identify metals and non-	the table to identify metals and non-metals.	Explain the properties of	purposes, based on their
Patterns	metals.	and non-metats.	elements using data and why	properties.
	inetats.	Identify metals and non-	they are used for different	properties.
	Understand what a	metals using data and	applications.	Make links between simple
	compound is.	suggest a reason for	applications.	models of compounds and
	Compound is.	applications.	Explain how compounds can be	chemical symbols.
	State what an ion is.	аррисацопъ.	formed and explain a chemical	chemical symbols.
	Guite What all loll is.		reaction using simple models.	
			reaction using simple models.	

Year 9 Science	Emerging	Developing	Secure	Mastery
		Describe an example of a compound and represent a chemical reaction using a simple model.  Describe where ionic bonding takes place.	Draw dot & cross diagrams to represent Ionic bonding.	